

# Larmor Precession

$$\vec{B} = B_0 \hat{z} \quad \vec{\mu} = \gamma \vec{S} \quad \gamma \approx -\frac{e}{m}$$

$$H = -\vec{\mu} \cdot \vec{B} = -\gamma B_0 S_z = \frac{\omega \hbar}{2} \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}$$

$$i\hbar \frac{\partial}{\partial t} \psi(t) = H \psi(t)$$

$$\psi(t) = \begin{pmatrix} \cos(\alpha/2) e^{-i\omega t/2} \\ \sin(\alpha/2) e^{+i\omega t/2} \end{pmatrix}$$

# Larmor Precession

$$\begin{aligned}\langle \psi | S_z | \psi \rangle &= \begin{pmatrix} \cos(\alpha/2)e^{i\omega t/2} & \sin(\alpha/2)e^{-i\omega t/2} \end{pmatrix} \frac{\hbar}{2} \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix} \\ &\quad \times \begin{pmatrix} \cos(\alpha/2)e^{-i\omega t/2} \\ \sin(\alpha/2)e^{i\omega t/2} \end{pmatrix} \\ &= \frac{\hbar}{2} (\cos^2(\alpha/2) - \sin^2(\alpha/2)) \\ &= \frac{\hbar}{2} \cos(\alpha)\end{aligned}$$

# Larmor Precession

$$\begin{aligned}\langle \psi | S_x | \psi \rangle &= \begin{pmatrix} \cos(\alpha/2)e^{i\omega t/2} & \sin(\alpha/2)e^{-i\omega t/2} \end{pmatrix} \frac{\hbar}{2} \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix} \\ &\quad \times \begin{pmatrix} \cos(\alpha/2)e^{-i\omega t/2} \\ \sin(\alpha/2)e^{i\omega t/2} \end{pmatrix} \\ &= \frac{\hbar}{2} \cos(\alpha/2) \sin(\alpha/2) (e^{i\omega t} + e^{-i\omega t}) \\ &= \frac{\hbar}{2} \sin(\alpha) \cos(\omega t)\end{aligned}$$

# Larmor Precession

$$\begin{aligned}\langle \psi | S_y | \psi \rangle &= \begin{pmatrix} \cos(\alpha/2)e^{i\omega t/2} & \sin(\alpha/2)e^{-i\omega t/2} \end{pmatrix} \frac{\hbar}{2} \begin{pmatrix} 0 & -i \\ i & 0 \end{pmatrix} \\ &\quad \times \begin{pmatrix} \cos(\alpha/2)e^{-i\omega t/2} \\ \sin(\alpha/2)e^{i\omega t/2} \end{pmatrix} \\ &= \frac{\hbar}{2} \cos(\alpha/2) \sin(\alpha/2) i (-e^{i\omega t} + e^{-i\omega t}) \\ &= \frac{\hbar}{2} \sin(\alpha) \sin(-\omega t)\end{aligned}$$

# Larmor Precession

$$\langle \psi | S_x | \psi \rangle = \frac{\hbar}{2} \sin(\alpha) \cos(\omega t)$$

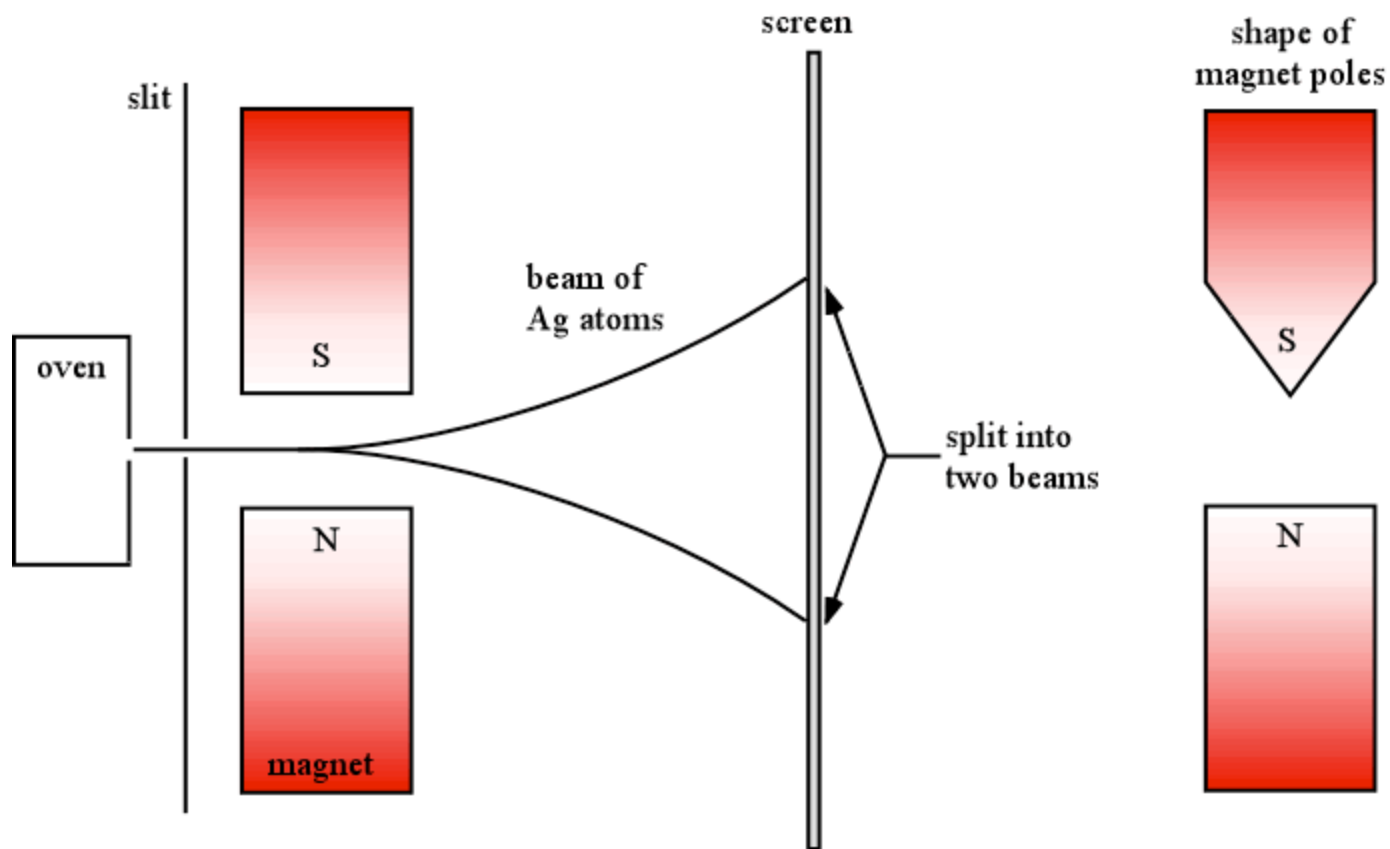
$$\langle \psi | S_y | \psi \rangle = \frac{\hbar}{2} \sin(\alpha) \sin(-\omega t)$$

$$\langle \psi | S_z | \psi \rangle = \frac{\hbar}{2} \cos(\alpha)$$

$$\omega = -\gamma B_0$$

# Stern-Gerlach

## Nobel Prize 1943



STERN-GERLACH EXPERIMENT

# Stern-Gerlach

$$H = -\gamma \vec{B} \cdot \vec{S}$$

$$\vec{B} = -\alpha x \hat{x} + (B_0 + \alpha z) \hat{z} \quad \vec{\nabla} \cdot \vec{B} = 0$$

silver atom rest frame,  $x=0$

$$H = \begin{cases} 0 & t < 0 \\ -\gamma(B_0 + \alpha z)S_z & 0 \leq t \leq T \\ 0 & T < t \end{cases}$$

$$\chi(t < 0) = a\chi_{\uparrow} + b\chi_{\downarrow}$$

$$\chi(0 < t < T) = a\chi_{\uparrow} e^{-i\omega t/2 + i\gamma\alpha z t/2} + b\chi_{\downarrow} e^{i\omega t/2 - i\gamma\alpha z t/2}$$

$$\omega = -\gamma B_0$$

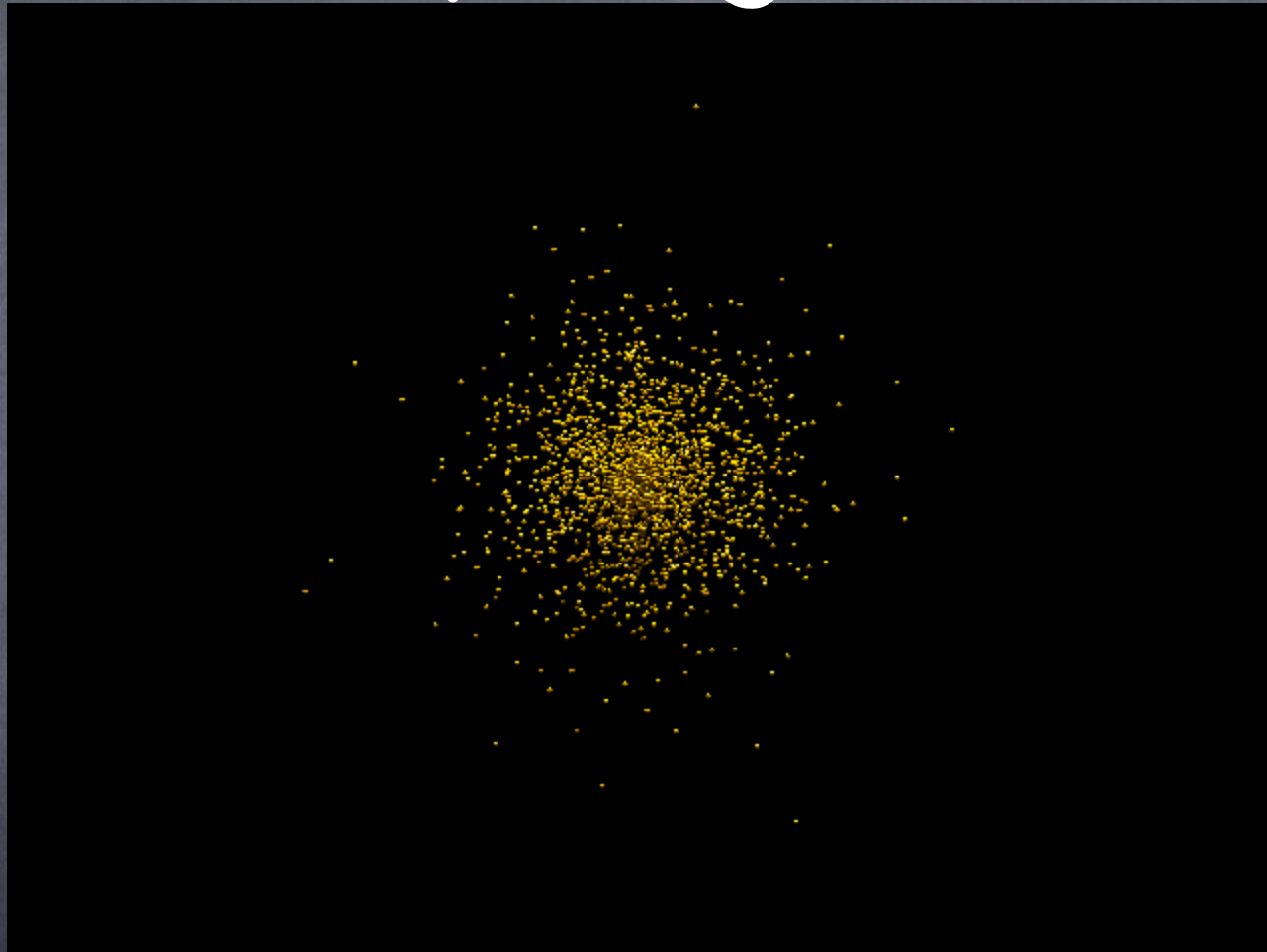
# Stern-Gerlach

$$\chi(t > T) = a\chi_{\uparrow}e^{-i\omega T/2+i\gamma\alpha zT/2} + b\chi_{\downarrow}e^{i\omega T/2-i\gamma\alpha zT/2}$$

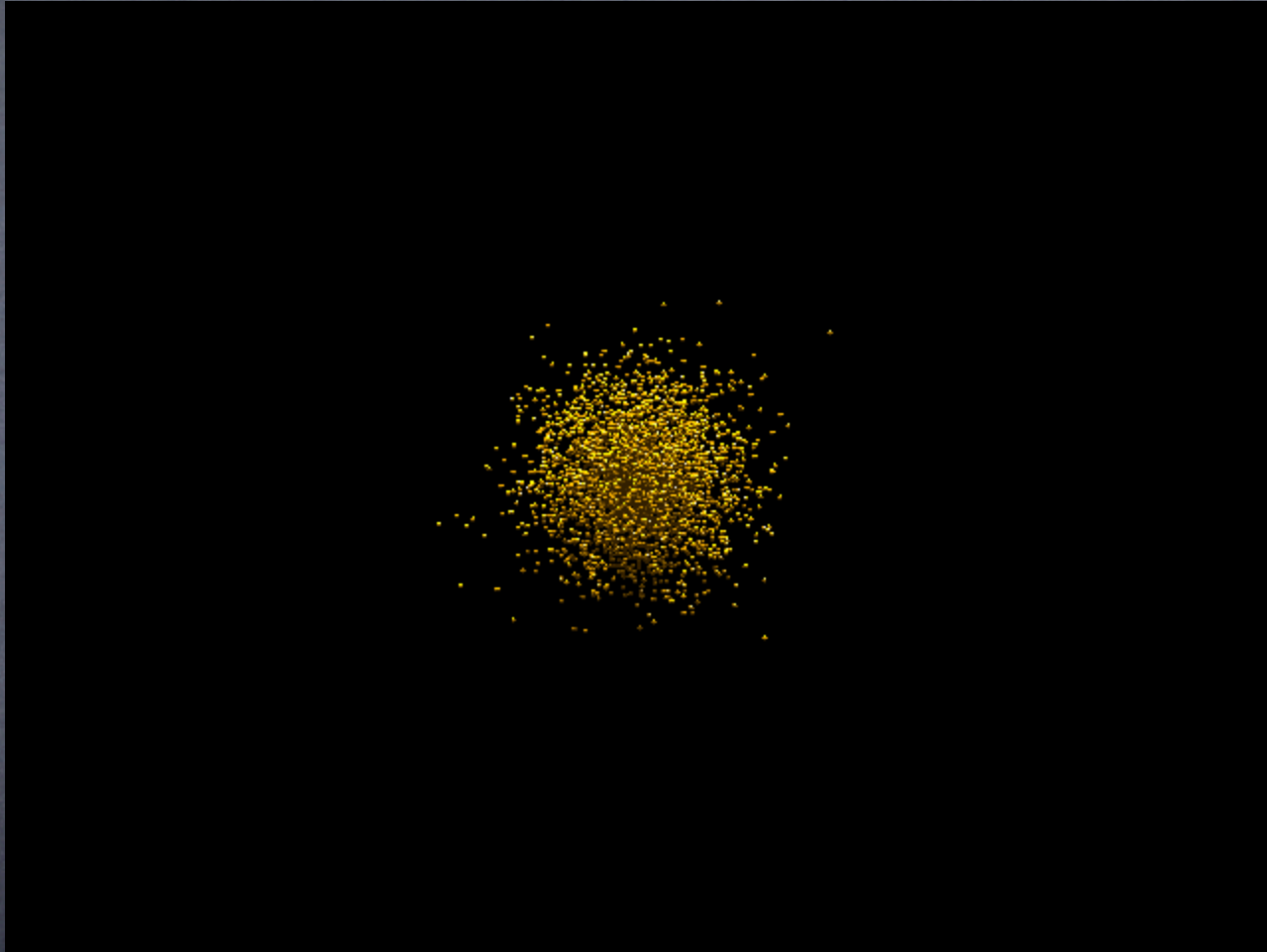
$$p_z = \pm \frac{\alpha\gamma T\hbar}{2}$$



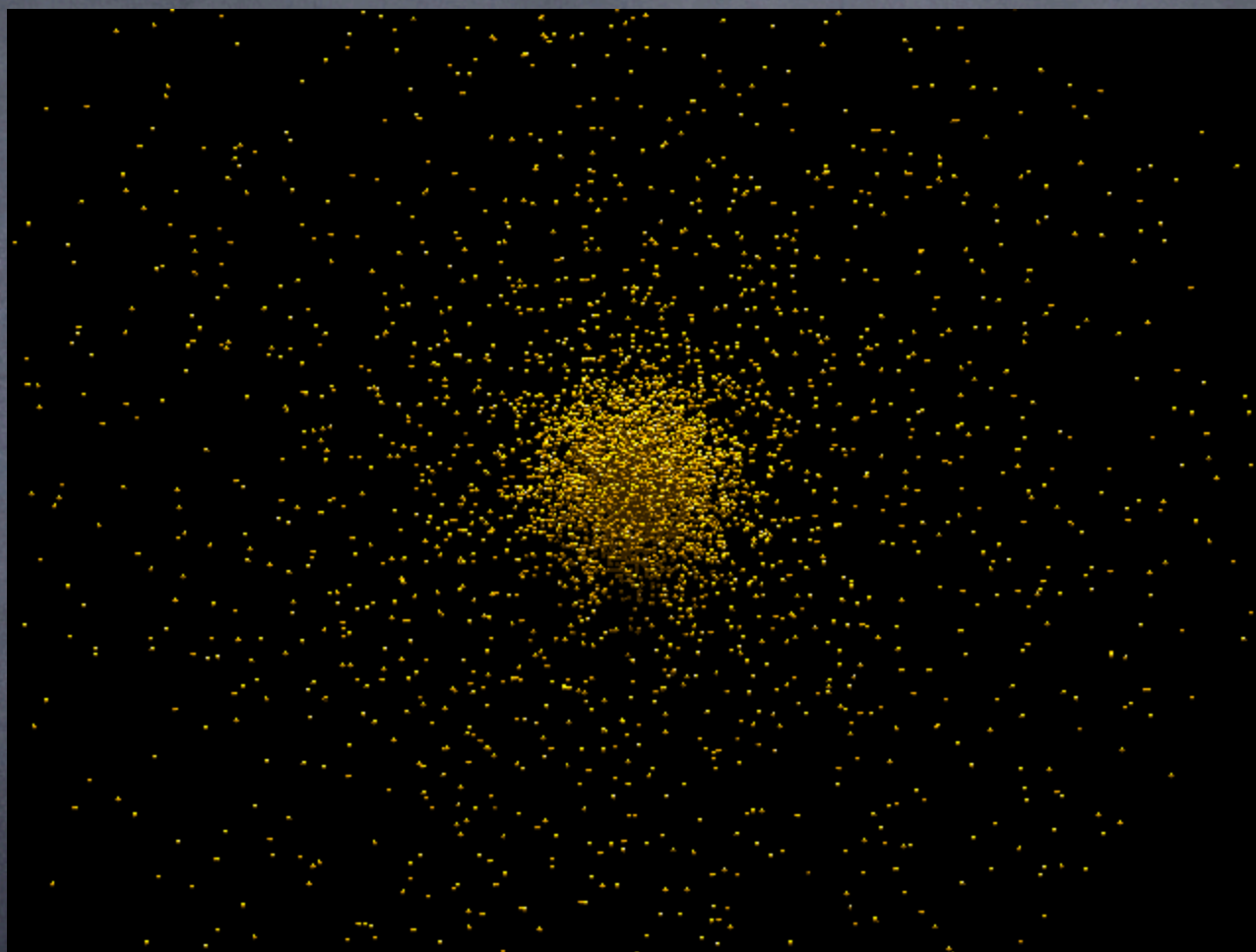
# Hydrogen



# Helium



# Lithium



# Elements

$Z$  protons

ignoring repulsion,  $(n, l, m)$  orbitals

2 electrons per orbital  $\uparrow\downarrow$  (spin singlet)

$n^2$ -fold degeneracy

$n = 1$     2 electrons

$n = 2$      $2^2 \cdot 2 = 8$  electrons

$n = 3$      $3^2 \cdot 2 = 18$  electrons

$n = 4$      $4^2 \cdot 2 = 32$  electrons

$n = 5$      $5^2 \cdot 2 = 50$  electrons

$\vdots$

$\vdots$



# Elements

Z	Element	outer electrons
1	Hydrogen	1 in (1,0,0)
2	Helium	2 in (1,0,0)
3	Lithium	1 in (2,0,0)
4	Beryllium	2 in (2,0,0)
5	Boron	1 in (2,1,m)
6	Carbon	2 in (2,1,m)
7	Nitrogen	3 in (2,1,m)
8	Oxygen	4 in (2,1,m)
9	Fluorine	5 in (2,1,m)
10	Neon	6 in (2,1,m)
11	Sodium	1 in (3,0,0)
12	Magnesium	2 in (3,0,0)
13	Aluminum	1 in (3,1,m)
14	Silicon	2 in (3,1,m)
15	Phosphorous	3 in (3,1,m)
16	Sulfur	4 in (3,1,m)
17	Chlorine	5 in (3,1,m)
18	Argon	6 in (3,1,m)
19	Potassium	1 in (4,0,0)
20	Calcium	2 in (4,0,0)
21	Scandium	1 in (3,2,m)
⋮	⋮	⋮

$$\left. \begin{array}{l} 5 \\ 4 \\ 3 \\ 2 \\ 1 \end{array} \right\} (2\ell+1) \times 2 = 6$$

$$\left. \begin{array}{l} 5 \\ 4 \\ 3 \\ 2 \\ 1 \end{array} \right\} (2\ell+1) \times 2 = 6$$

# Elements

$2s+1L_J$

36 Kr Krypton  $[\text{Ar}](4s)^2(3d)^{10}(4p)^6$   $^1S_0$

47 Ag Silver  $[\text{Kr}](4d)^{10}(5s)^1$   $^2S_{1/2}$